

FLAVONOIDS FROM *GUTIERREZIA REPENS* (ASTERACEAE)

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Received September 18, 2007. In final form November 23, 2007

Abstract

7,3'-dimethylquercetin **1**, 7,3,3'-trimethylquercetin **2**, 7,3,4'-trimethylquercetin **3** and quercetin **4** were isolated from aerial parts of *Gutierrezia repens* (Asteraceae). The structures of **1**, **2** and **3** were determined mainly on the basis of 2D NMR data. Their ¹H NMR spectra in CDCl₃ and Me₂CO-d₆ are compared and discussed. The ¹³C NMR spectra of these compounds are given here for the first time.

Keywords: *Gutierrezia repens*, Asteraceae, phytochemistry, flavonoids

Resumen

7,3'-dimetilqueracetina **1**, 7,3,3'-trimetilqueracetina **2**, 7,3,4'-trimetilqueracetina **3** y queracetina **4** fueron aislados de las partes aéreas de *Gutierrezia repens* (Asteraceae). Las estructuras de **1**, **2** y **3** fueron determinadas principalmente por espectroscopía 2D RMN. Sus espectros de RMN ¹H en CDCl₃ y Me₂CO-d₆ son comparados y discutidos. En este trabajo informamos por primera vez, los espectros de RMN ¹³C de los flavonoides metilados.

Palabras clave: *Gutierrezia repens*, Asteraceae, fitoquímica, flavonoides.

Introduction

The Asteraceae is the second largest family in the Magnoliophyta Division with around 1100 genera and over 20000 recognized species. Cabrera, reported the occurrence of 197 genera and about 1400 species in Argentina [1].

As part of our phytochemical study on Asteraceae species growing in Argentina, we investigated the aerial parts of *Gutierrezia repens* Grisebach. There is no information about chemical and biological studies carried out on *G. repens*. Plant specimens were collected from their natural habitat in the northwest of Argentina, in Salta Province.

The genus *Gutierrezia* (tribe Eupatorieae) includes approximately 25 species which occur exclusively in the arid areas of America [1]. Earlier work on this genus revealed that diterpenes [2-8] and flavonoids [9-15] are the main classes of substances representative of the *Gutierrezia* genus.

In this paper, we report for the first time on a phytochemical investigation of *G. repens*.

Experimental

General

The NMR spectra were recorded on a Bruker AC 200 (¹H at 200 MHz and ¹³C at 50 MHz) or a Bruker Avance 400 (¹H at 400 MHz and ¹³C at 100 MHz) spectrometer with TMS as internal reference. CC were performed on silica-gel 230-400 mesh, RPCC on C-18 silica gel, TLC was carried out on precoated Silica gel 60 F₂₅₄ plates (Fluka). Detection was achieved by UV light and spraying with vanillin reagent followed by heating.

Plant Material

G. repens was collected during the flowering period in Valle Encantado, Province of Salta, Argentina, on February 2004. The identification was carried out by Ing. Julio Tolaba. A voucher specimen (nº 3464) is deposited at the Museo de la Facultad de Ciencias Naturales, Universidad Nacional de Salta.

Extraction and isolation

Air-dried and powdered aerial parts of *G. repens* (260 g) were macerated with EtOH at room temperature for 7 days to give 13.10 g of crude extract which was suspended in EtOH:H₂O (1:1) and extracted successively with hexane (3x150 mL), CH₂Cl₂ (3x150 mL) and EtOAc (3x100 mL). Evaporation of the CH₂Cl₂ extract in vacuo furnished 5.37 g of residue which was divided into 3 fractions by chromatography on reversed-phase silica gel flash column, eluting with MeOH-H₂O (8:2), MeOH and Me₂CO. The fraction 1 (2.0 g) was chromatographed on a 230-400 mesh silica gel column using hexane containing increasing amounts of EtOAc (0-100 %), seven fractions being collected (F₁ to F₇). Fraction F₅ (269 mg, hexane-EtOAc 3:7), was first purified by column chromatography on silica gel eluting with a gradient of hexane-Et₂O followed by preparative TLC (hexane-Me₂CO 7:3) affording 2.5 mg of 7,3'-dimethylquercetin 1 (R_f= 0.30) [16], 3.0 mg of 7,3,3'-trimethylquercetin 2 (R_f= 0.36) [16, 17] and 3.5 mg of 7,3,4'-trimethylquercetin 3 (R_f= 0.33) [18]. Column chromatography of Fraction F₆ (230 mg, hexane-EtOAc 1:9) on silica gel and benzene-EtOAc gradient system followed by preparative TLC (hexane-Me₂CO, 1:1) afforded 7.0 mg of quercetin 4 (R_f= 0.48) [19].

7,3'-dimethylquercetin 1. Amorphous solid, UV (MeOH) λ_{\max} nm: 260, 270, 370; +NaOMe: 270, 300, 330, 430 (dec); +NaOAc: 260, 335, 375; +AlCl₃: 275, 430; +AlCl₃/HCl: 275, 430.

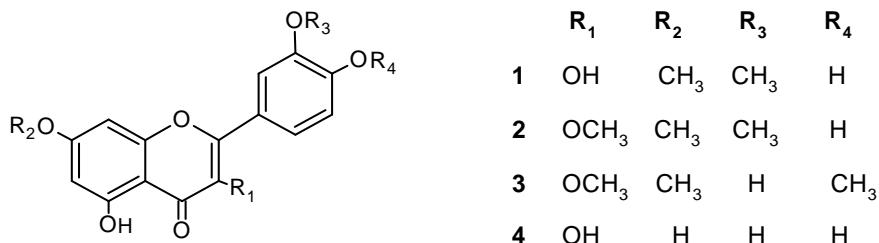
7,3,3'-trimethylquercetin 2. Amorphous solid, UV (MeOH) λ_{\max} nm: 270, 348, 355; +NaOMe: 265, 405; +NaOAc: 270, 355; +AlCl₃: 270, 302, 406; +AlCl₃/HCl: 348, 406.

7,3,4'-trimethylquercetin 3. Amorphous solid, UV (MeOH) λ_{\max} nm: 270, 300, 348; +NaOMe: 265, 375; +NaOAc: 265, 355; +AlCl₃: 270, 300, 362, 400; +AlCl₃/HCl: 270, 300, 362, 400.

quercetin 4. Yellow solid, UV (MeOH) λ_{\max} nm: 255, 300, 370, 385; +NaOMe: 330 sh; +NaOAc: 274, 394; +AlCl₃: 266, 300, 358, 430; +AlCl₃/HCl: 266, 300, 358, 430.

Discussion

The CH_2Cl_2 soluble extract of the aerial parts of *G. repens* Griseb. yielded four known flavonoids 7,3'-dimethylquercetin **1** [16], 7,3,3'-trimethylquercetin **2** [16, 17], 7,3,4'-trimethylquercetin **3** [18] and quercetin **4** [19].



Bathochromic shifts upon addition of AlCl_3 and AlCl_3/HCl (see experimental) together with the presence of a chelated hydroxyl group in the ^1H NMR spectrum (Table 1 and Table 2), indicated 5-hydroxy substitution for all four compounds.

Table 1. Spectroscopic data of flavonoid **1** and **4*** ($\text{Me}_2\text{CO-d}_6$, TMS as internal standard).

Position	1				4
	δ (C)*	δ (H)*	δ (H) [†]	HMBC*	
2	146.5	-	-	-	-
3	135.3	-	-	-	-
4	175.0	-	-	-	-
5	161.1	-	-	-	-
6	97.4	6.33 d (2.0)	6.36 (2.2)	C-5, C-7, C-10	6.27 d (2.2)
7	165.8	-	-	-	-
8	92.0	6.80 d (2.0)	6.48 d (2.2)	C-7, C-9, C-10	6.53 d (2.2)
9	156.9	-	-	-	-
10	103.9	-	-	-	-
1'	121.9	-	-	-	-
2'	111.2	7.90 d (2.2)	7.7-7.8 m	C-1', C-4'	7.82 d(2.0)
3'	147.5	-	-	-	-
4'	148.9	-	-	-	-
5'	116.0	7.02 d (8.5)	7.02 d (8.8)	C-6',C-3'	7.00 d (8.4)
6'	122.6	7.86 dd (8.5, 2.2)	7.7-7.8 m	C-4', C-2'	7.70 dd (8.4, 2.0)
3'-OCH₃	55.6	3.95 s	3.97 s	C-3'	-
7-OCH₃	55.5	3.94 s	3.86 s	C-7	-
5-OH		12.14 s	12.62 s		12.60 s

* At 400 MHz for ^1H NMR and 100 MHz for ^{13}C NMR.

† At 200 MHz in CDCl_3 .

δ (H) values are followed by multiplicity and below, in parentheses, coupling constants in Hz.

Table 2. Spectroscopic data of flavonoid **2** and **3*** ($\text{Me}_2\text{CO-d}_6$, TMS as internal standard)

Position	2				3			
	δ (C)*	δ (H)*	δ (H)†	HMBC*	δ (C)*	δ (H)*	δ (H)†	HMBC*
2	155.9	-	-	-	155.9	-	-	-
3	138.6	-	-	-	138.9	-	-	-
4	178.7	-	-	-	178.5	-	-	-
5	162.0	-	-	-	162.0	-	-	-
6	97.3	6.33 d (2.0)	6.37 d (2.0)	C-5, C-7, C-10	97.7	6.33 d (2.0)	6.36 d (2.2)	C-5, C-7, C-10
7	165.7	-	-	-	165.9	-	-	-
8	91.8	6.68 d (2.0)	6.45 d (2.0)	C-9	92.0	6.70 d (2.0)	6.45 d (2.2)	C-7, C-9, C-10
9	156.8	-	-	-	156.9	-	-	-
10	105.7	-	-	-	105.7	-	-	-
1'	121.9	-	-	-	123.2	-	-	-
2'	111.8	7.80 d (2.0)	7.71 s, br ‡	C-3'	114.9	7.67 d ‡ (2.2)	7.70 d ‡ (2.0)	C-3', C-4', C-6'
3'	147.4	-	-	-	146.4	-	-	-
4'	149.7	-	-	-	150.1	-	-	-
5'	115.2	7.02 d (8.5)	7.05 d (8.3)	C-4'	111.2	7.15 d (8)	6.97 d (8.3)	C-4', C-1'
6'	122.5	7.72 dd (8.5, 2.2)	7.67 dd ‡ (8.3, 2.2)	C-4'	121.0	7.72 dd ‡ (8.0, 2.2)	7.72 dd ‡ (8.0, 2.2)	-
OCH₃	55.5	3.95 s	3.98 s	C-3'	55.4	3.97 s	3.99 s	C-4'
7-OCH₃	55.5	3.93 s	3.87 s	C-7	55.4	3.94 s	3.88 s	C-7
3-OCH₃	59.1	3.90 s	3.86 s	C-3	59.3	3.90 s	3.88 s	C-3
5-OH		12.73 s	12.63 s	C-5		12.70 s	12.70 s	C-5

* At 400 MHz for ^1H NMR and 100 MHz for ^{13}C NMR.† At 200 MHz in CDCl_3 . δ (H) values are followed by multiplicity and below, in parentheses, coupling constants in Hz.

‡ Overlapped signals.

In the ^1H NMR spectra of all the compounds three aromatic protons formed the characteristic pattern for a 3',4'-disubstituted B ring. Additionally, the UV spectra recorded with NaOMe indicated 4'-hydroxy substitution for **1**, **2** and **4**.

Flavonoids **1**, **2** and **3** also showed ^1H NMR signals indicative of O-methyl substituents. Their UV spectra were unchanged upon addition of NaOAc, indicating that one of the methoxyl groups was at the C-7 position. The structures of these compounds were deduced on the basis of their HSQC, HMBC and NOESY spectra.

The ^1H NMR data of known compounds **1**, **2** and **3** were previously measured using low resolution instrument. As far as we know, the ^{13}C NMR spectra of flavonoids **1**, **2** and **3** have not been described in the literature so far (Tables 1 and 2).

The ^1H NMR data of **3**, indicate that 4'-O-methylation induces a downfield shift of ca. 0.15 ppm in the signal of H-5', in the spectrum measured in $\text{Me}_2\text{CO-d}_6$ (Table 2). In the spectrum measured in CDCl_3 this effect is clearly smaller (less than 0.1 ppm) (Table 2). On the other hand, in all the compounds with 7-O-methylation (**1**, **2** and **3**), we always observed a downfield shift of 0.15-0.30 ppm in the signal of H-8, in spectra measured in $\text{Me}_2\text{CO-d}_6$ (Tables 1 and 2).

The A-Ring carbon signals are similar in **1**, **2** and **3**. B-ring signals show that 4'-O-methylation in **3** induces an upfield shift of ca. 4.0 ppm in the chemical resonance of C-5' (Table 2).

Conclusions

7,3'-dimethylquercetin **1** and 7,3,3'-trimehtylquercetin **2** are now reported for the first time in the genus, while 7,3,4'-trimethylquercetin **3** was isolated before from *G. alamanii* [13] and quercetin **4** from *G. grandis* [12], *G. alamanii* [13], *G. wrightii* [14] and *G. microcephala* [15].

The isolation of compounds **1-4** from *G. repens* is completely in accordance with the typical chemical profile of the *Gutierrezia* genus.

Acknowledgments

Thanks are due to the Consejo de Investigación de la Universidad Nacional de Salta for financial supports.

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